

# CMC carboxymethyl cellulose

PARAMETER	UNIT	VALUE	REFERENCES
<b>GENERAL</b>			
Common name	-	carboxymethyl cellulose	
IUPAC name	-	acetic acid; 2,3,4,5,6-pentahydroxyhexanal	
ACS name	-	cellulose, carboxymethyl ether	
Acronym	-	CMC	
CAS number	-	9000-11-7	
RTECS number	-	FJ5700000	
<b>HISTORY</b>			
Person to discover	-	Payen, A	
Date	-	1838	
Details	-	Payen determined elemental composition of carboxymethyl cellulose in 1838	
<b>SYNTHESIS</b>			
Monomer(s) structure	-	chloroacetic acid	
Monomer(s) CAS number(s)	-	79-11-8	
Monomer(s) molecular weight(s)	dalton, g/mol, amu	94.50	
Degree of substitution	-	0.4-1.5 (theoretical maximum is 3 when all 3 groups in monomeric unit are substituted); 0.11-2.41	Yeasmin, S; Mondal, I H, Int. J. Biol. Macromol., 80, 725-31, 2015
Method of synthesis	-	carboxymethyl cellulose is obtained from reaction between cellulose and chloroacetic acid in the presence of alkalis which catalyze reaction	
Catalyst	-	alkalis	
Mass average molecular weight, $M_w$	dalton, g/mol, amu	80,000-560,000	
Polymerization degree (number of monomer units)	-	350-2,500	
<b>STRUCTURE</b>			
Crystallinity	%	80	Li, H; Wu, B; Mu, C; Lin, W, Carbohydrate Polym., 84, 881-86, 2011.
<b>COMMERCIAL POLYMERS</b>			
Some manufacturers	-	Dow	
Trade names	-	Walocel	
<b>PHYSICAL PROPERTIES</b>			
Density at 20°C	g cm <sup>-3</sup>	1.05	
Decomposition temperature	°C	140-150	
pH	-	6.8	Yeasmin, S; Mondal, I H, Int. J. Biol. Macromol., 80, 725-31, 2015
Permeability to water vapor, 25°C	g mm m <sup>-2</sup> day <sup>-1</sup> kPa <sup>-1</sup>	1.8	Sayanjali, S; Ghanbarzadeh, B; Ghiassifar, S, LWT Food Sci. Technol., 44, 1133-38, 2011.

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<b>MECHANICAL &amp; RHEOLOGICAL PROPERTIES</b>			
Tensile strength	MPa	17.6-17.8	Sayanjali, S; Ghanbarzadeh, B; Ghiassifar, S, LWT Food Sci. Technol., 44, 1133-38, 2011.
Elastic modulus	MPa	1,350	Ghanbarzadeh, B; Almasi, H, Int. J. Biol. Macromol., 48, 44-49, 2011.
Young's modulus	MPa	1227	Su, J-F; Huang, Z; Yuan, X-Y; Wang, X-Y; Li, M, Carbohydrate Polym., 79, 145-53, 2010.
Water absorption, equilibrium in water at 23°C	%	6.5; 10 max.	
<b>CHEMICAL RESISTANCE</b>			
Acid dilute/concentrated	-	poor	
Alcohols	-	poor	
Alkalis	-	poor	
Aliphatic hydrocarbons	-	good	
Aromatic hydrocarbons	-	fair	
Esters	-	poor	
Greases & oils	-	fair	
Halogenated hydrocarbons	-	poor	
Ketones	-	poor	
Good solvent	-	alkalies, acetone, chloroform, esters, mixture of water and alcohols, pyridine, water	
<b>FLAMMABILITY</b>			
Autoignition temperature	°C	287-370	
<b>BIODEGRADATION</b>			
Typical biodegradants	-	bacteria which can produce cellulase	
Stabilizers	-	Carbosan; potassium sorbate	Sayanjali, S; Ghanbarzadeh, B; Ghiassifar, S, LWT Food Sci. Technol., 44, 1133-38, 2011.
<b>TOXICITY</b>			
NFPA: Health, Flammability, Reactivity rating	-	0/2/0	
Carcinogenic effect	-	not listed by ACGIH, NIOSH, NTP	
Oral rat, LD <sub>50</sub>	mg kg <sup>-1</sup>	>5,000	
Skin rabbit, LD <sub>50</sub>	mg kg <sup>-1</sup>	>5,000	
<b>PROCESSING</b>			
Typical processing methods	-	compounding	
Applications	-	ceramics, cosmetics, fabric finishing, food products, flotation, leather pasting, paints, paper, pharmaceuticals, textile sizing, thickener and emulsion stabilizer in coatings, toothpaste, washing powders and liquids, well drilling	

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<b>BLENDS</b>			
Suitable polymers	-	PEO, PR, PAM, PANI, carrageenan, PVAI	Miao, J; Zhang, R; Bai, R, J. Membrane Sci., 493, 654-63, 2015.
<b>ANALYSIS</b>			
FTIR (wavenumber-assignment)	cm <sup>-1</sup> /-	3420 - OH stretching, 2925 - CH and CO stretching, 1640 - H-O-H, C=C, 896 - glucosidic linkages	Yeasmin, S; Mondal, I H, Int. J. Biol. Macromol., 80, 725-31, 2015.

# POLYMER PART

